

Methodology to determine carbonate carbon from Thermal Optical Transmittance measurements

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EAC2011 Manchester European Aerosol Conference 4-9 Sept 2011

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Carbonate carbon (CC) is often not considered in atmospheric aerosol chemistry studies which comprise the measurement of elemental carbon (EC) and organic carbon (OC). The reason for this may be its low contribution to fine particle mass in most area along with the difficulties in its analytical determination in atmospheric aerosol collected on filter matrices.

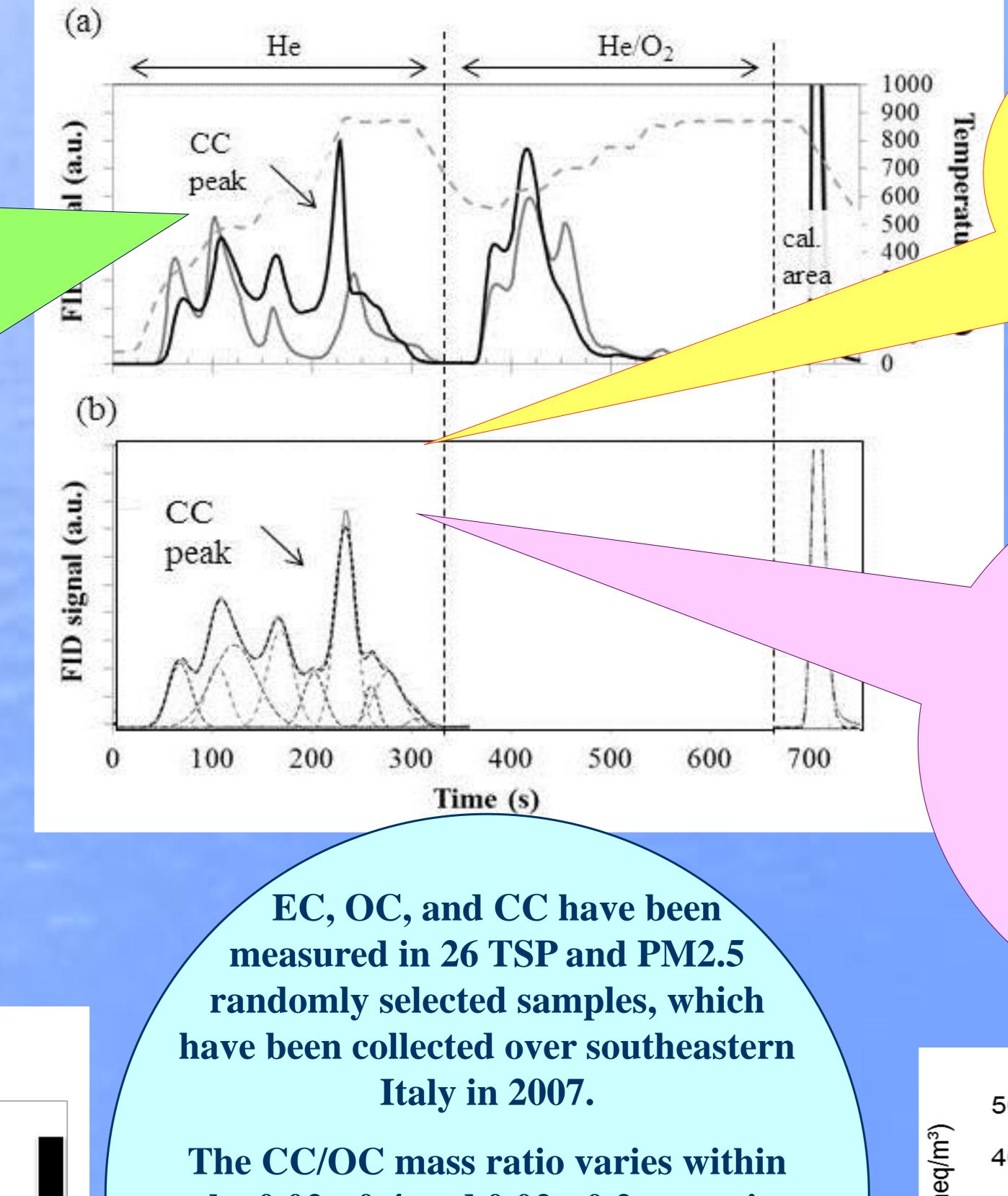
However, the CC fraction in particulate matter may not be negligible if high concentrations of mineral dust, either natural (natural erosion, sand storms) or originating from street abrasion or construction sites are present.

We have implemented a methodology to determine CC, OC and EC levels from Thermal Optical Transmittance (TOT) measurements on ground collected samples without pre treating the samples with HCl fumes.

Methodology

We have used the sample pretreatment with HCl fumes to identify the CC contribution to the flame ionization detector (FID) signal from TOT measurements.

The time evolution analysis of the FID signal before (black line) and after (grey line) the treatment with HCl fumes, has revealed that the CC peak may



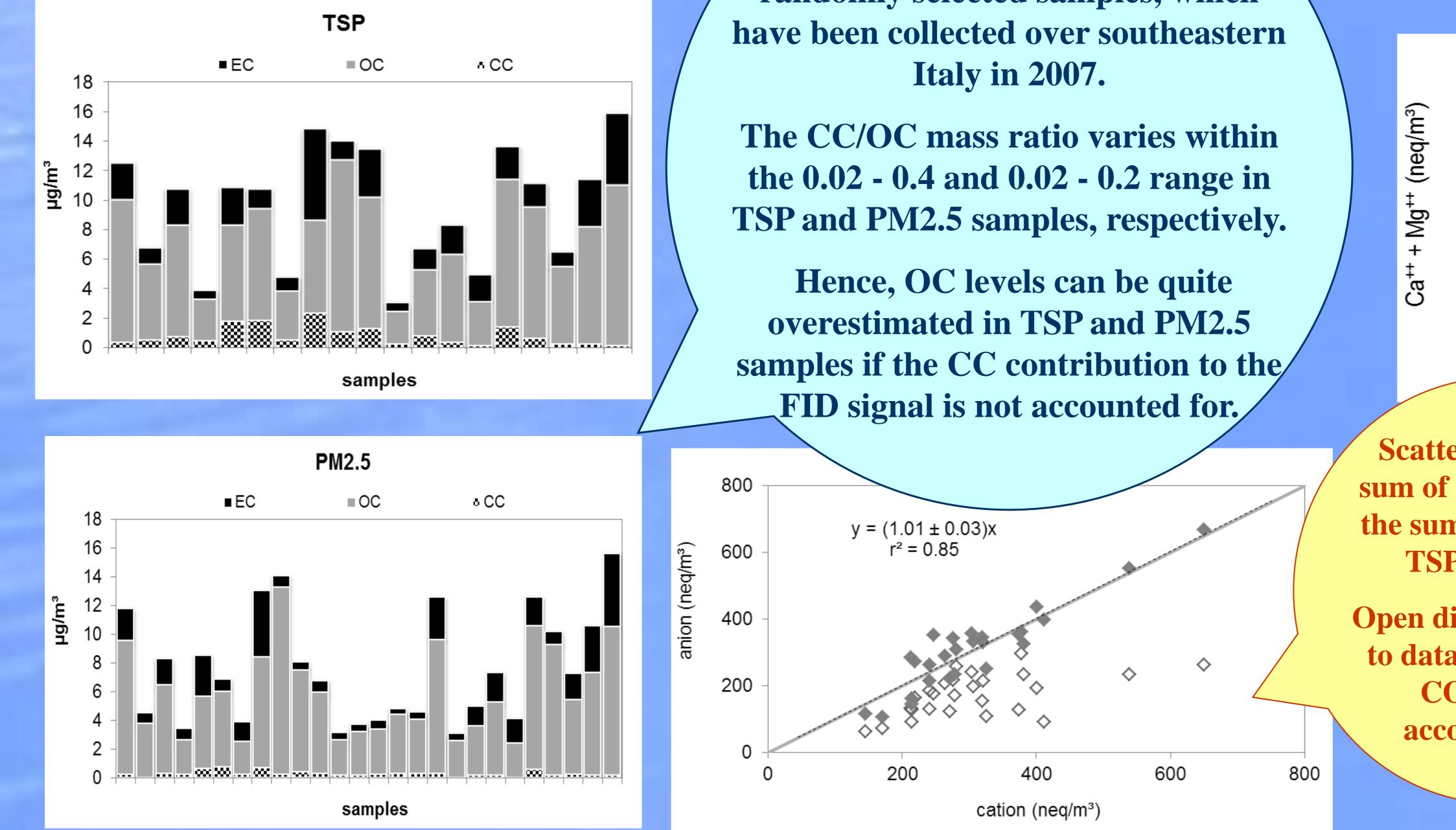
The FID signal has been represented as a weighted sum of Gaussian functions:

 $S(t) = \sum_{i} a_{i} \exp[-(t-t_{i})^{2}/2\sigma_{i}^{2}]$ 

where a<sub>i</sub>, t<sub>i</sub>, and σ<sub>i</sub> represent amplitude, peak-time and standard deviation of the Gaussian function i.

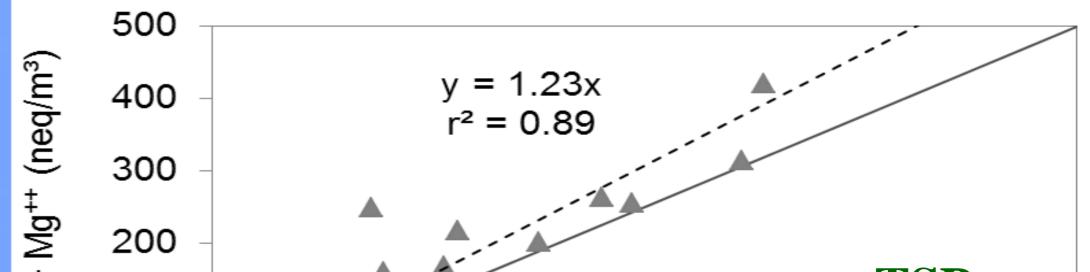
occurs within the 220-250 s time interval and that it is characterized by a full-with-at-halfmaximum  $\Delta t^* = 25 \pm 3$  s

Results



We have assumed that the CC volatilization contributes to the FID signal with a pulse which can be fitted by a Gaussian function with the peak at a time  $t_i$  (within the 220-250 s time interval) and a full-width-at-halfmaximum  $\Delta t_i = 25\pm 3$  s.

Then, we have calculated the Gaussian function area ascribed to the CC volatilization and the area of the calibration signal, to quantify CC levels.



TSP 100 500 200 300 400 100 CO<sub>3</sub>-- (neq/m<sup>3</sup>) **Scatter plot of the** CO<sub>3</sub><sup>2-</sup> is mainly neutralized by sum of anions versus  $Ca^{2+}$  and  $Mg^{2+}$ . the sum of cations in **TSP** samples. Acknowledgements **Open diamonds refer** Dr. P. Fermo is acknowledged for her to data points when scientific support. Most of the chemical  $CO_3^{2-}$  is not analyses have been carried out at the accounted for. **Department of Chimica Inorganica**, Metallorganica e Analitica, University of Milan (Italy).